This article was downloaded by:

On: 26 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-

41 Mortimer Street, London W1T 3JH, UK



Nucleosides, Nucleotides and Nucleic Acids

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713597286

A NEW PROCEDURE FOR THE PHOSPHORYLATION OF NUCLEOSIDES: APPLICATION TO THE DISCOVERY OF INHIBITORS OF HIV INTEGRASE

Vasu Naira; Michael Taktakishvilia

^a Department of Chemistry, The University of Iowa, Iowa City, Iowa, U.S.A.

Online publication date: 31 March 2001

To cite this Article Nair, Vasu and Taktakishvili, Michael (2001) 'A NEW PROCEDURE FOR THE PHOSPHORYLATION OF NUCLEOSIDES: APPLICATION TO THE DISCOVERY OF INHIBITORS OF HIV INTEGRASE', Nucleosides, Nucleotides and Nucleic Acids, 20: 4, 739 - 742

To link to this Article: DOI: 10.1081/NCN-100002419 URL: http://dx.doi.org/10.1081/NCN-100002419

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

A NEW PROCEDURE FOR THE PHOSPHORYLATION OF NUCLEOSIDES: APPLICATION TO THE DISCOVERY OF INHIBITORS OF HIV INTEGRASE

Vasu Nair* and Michael Taktakishvili

Department of Chemistry, The University of Iowa, Iowa City, Iowa 52242

ABSTRACT

A new phosphorylating agent for nucleosides, 2-O-(4,4'-dimethoxytrityl) ethylsulfonylethan-2'-yl-phosphate (1), has been developed by us. In the many examples studied by us, phosphorylation yields were found to be very high (about 90%). The procedure appears to be remarkably general and can be utilized for the phosphorylation of many biomolecules. Successful application of this phosphorylation method has contributed to the discovery of inhibitors of HIV integrase in our laboratory.

INTRODUCTION

Phosphorylation reactions are ubiquitous in nature. Non-natural phosphorylated nucleosides are of interest for biological studies involving various nucleotide metabolizing enzymes and as potential therapeutic agents (1–4). Chemical methodologies for solution-phase phosphorylations are of critical importance in the preparation of these compounds. Several requirements need to be fulfilled for efficient phosphorylation of nucleosides. The hydroxyl functions of the phosphate group being introduced need to be protected (4). These phosphate protecting groups must possess stability with respect to reaction conditions of subsequent synthetic steps but must also have the property of selective lability at the termination of synthesis.

^{*}Corresponding author.

740 NAIR AND TAKTAKISHVILI

Scheme 1.

Current methods of chemical phosphorylation have disadvantages that range from lengthy procedures, instability of intermediates, difficulty of purification, problems associated with removal of phosphate protecting groups, issues of monitoring of reaction, and, in many cases low yields in the phosphorylation and deprotection steps (5–18).

We describe herein a new method for phosphorylation of nucleosides and nucleotides which utilizes more effective phosphodiester instead of phosphotriester chemistry and surmounts the problems associated with stability, lability, purification and yields in phosphorylation reactions discussed above. The new phosphorylating reagent is 2'-O-(4,4'-dimethoxytrityl)ethylsulfonylethan-2-yl-phosphate (1). It can be synthesized easily from sulfonyldiethanol by selective monotritylation followed by phosphorylation with POCl₃/1,2,4-triazole/triethylamine and work up with NaHCO₃ (Scheme 1) (19). When 1 is activated by coupling reagents such as TPS-TAZ or TPS-NT, it phosphorylates primary and secondary alcohol groups in nucleosides with remarkable efficiency (of the order of 90%). Deprotection of the phosphodiester intermediate formed after the phosphorylation step is carried out in only one step, but that step constitutes the normal work-up of the reaction. The precursor nucleoside does require protection prior to the phosphorylation step. An example is the 3'-phosphorylation of protected deoxyadenosine which proceeds in 90% overall yield under the conditions shown in Scheme 2.

We have examined many examples of 3'- as well as 5'-phosphorylation of nucleosides to establish the generality of this procedure (20). The results are completely reproducible. The phosphorylation methodology can be utilized in the synthesis of more complex and sensitive molecules, including those that are potent inhibitors of HIV integrase. An example of the utilization of this phosphorylation

Scheme 2.



Copyright @ Marcel Dekker, Inc. All rights reserved

PHOSPHORYLATION OF NUCLEOSIDES

Scheme 3.

method in the synthesis of one HIV integrase inhibitor is illustrated in Scheme 3. The yield of phosphorylated dinucleotide (HPLC purified) was 79%.

ACKNOWLEDGMENT

We thank the NIH (NIAID) for support of our research.

REFERENCES

- 1. Johnson, S.C.; Gerber, J.G. Advances in HIV/AIDS therapy. In "Advances in Internal Medicine," Volume 44. R.W. Schrier, J.D. Baxter, V.J. Dzau, A.S. Fauci, Eds. Mosby: St. Louis, **2000**, pp. 1–40.
- 2. DeClercq, E. Clin. Microbiol. Rev. 1997, 10, 674–693.
- 3. Hutchinson, D.W. In *Chemistry of Nucleosides and Nucleotides*, Townsend, L.B., Ed.; Plenum Press: New York, 1991, Vol. 2, pp 81–160.
- 4. Taktakishvili, M.; Neamati, N.; Pommier, Y.; Pal, S.; Nair, V. *J. Am. Chem. Soc.* **2000**, *122*, 5671–5677.
- 5. Takaku, H.; In *Nucleic Acid Chemistry*; Townsend, L.B.; Tipson, R.S., Ed.; Wiley: New York, 1978; Part 2, pp 865–874.
- 6. Zhang, H.-Z.; Rao, K.; Carr, S.F.; Straub, K.; Wu, C.J.; Fried, J. *J. Med. Chem.* **1997**, *40*, 4–8.
- Kang, S.H.; Sinhababu, A.K.; Cho, M.J. Nucleosides & Nucleotides, 1998, 17, 1089– 1098
- 8. Qui, X.; Ong, S.; Bernal, C.; Rhee, D.; Pidgeon, C. J. Org. Chem. 1994, 59, 537–547.
- 9. van der Marel, G.A.; van Boeckel, C.A.A.; Wille, G.; van Boom, J.H. *Nucleic Acids Res.* **1982**, *10*, 2337–2351.
- 10. Takaku, H.; Shimanda, Y.; Arai, K. Bull. Chem. Soc. Japan 1974, 47, 779–780.
- 11. Claesen, C.A.; Segers, R.P.; Tesser, G.I. Rec. Trav. Chim. Pays-Bas 1985, 104, 119–122.
- 12. Horn, T.; Urdea, M.S. Tetrahedron Lett. 1986, 27, 4705-4708.
- 13. Tanaka, T.; Yamada, Y.; Ikehara, M. Tetrahedron Lett. 1986, 27, 5641-5644.
- 14. Uhlman, E.; Engels, J. Tetrahedron Lett. 1986, 27, 1023–1026.



NAIR AND TAKTAKISHVILI

- 15. Ferris, J.P.; Rayser, J.R. Nucleosides & Nucleotides 1994, 13, 1087–1111.
- 16. Shaller, H.; Khorana, H.G. J. Am. Chem. Soc. 1963, 85, 3828–3835.
- 17. Nadeau, J.G.; Singleton, C.K.; Kelly, G.B.; Weith, H.L.; Gough, G.R. *Biochemistry*, **1984**, *23*, 6153–6159.
- 18. Rommler, D.H.; Lapidot, Y.; Khorana, H.G. J. Am. Chem. Soc. 1963, 85, 1989–1997.
- 19. Data for 1: 1 H NMR (CDCl₃): δ 7.45–7.25 (m, 9H, Arom), 6.35–6.30 (m, 4H, Arom), 4.20–4.10 (m, 2H, DMTrOCH₂), 3.65–3.45 (m, 2H, CH₂OP), 3.20–3.15 (m, 2H, DMTrOCH₂C<u>H₂</u>), 3.10–3.05 (m, 2H, C<u>H₂CH₂OP</u>). 31 P (CDCl₃): δ 4.90 (br,s).
- 20. Taktakishvili, M.; Nair, V. Tetrahedron Lett. 2000, In Press.

742

Request Permission or Order Reprints Instantly!

Interested in copying and sharing this article? In most cases, U.S. Copyright Law requires that you get permission from the article's rightsholder before using copyrighted content.

All information and materials found in this article, including but not limited to text, trademarks, patents, logos, graphics and images (the "Materials"), are the copyrighted works and other forms of intellectual property of Marcel Dekker, Inc., or its licensors. All rights not expressly granted are reserved.

Get permission to lawfully reproduce and distribute the Materials or order reprints quickly and painlessly. Simply click on the "Request Permission/Reprints Here" link below and follow the instructions. Visit the U.S. Copyright Office for information on Fair Use limitations of U.S. copyright law. Please refer to The Association of American Publishers' (AAP) website for guidelines on Fair Use in the Classroom.

The Materials are for your personal use only and cannot be reformatted, reposted, resold or distributed by electronic means or otherwise without permission from Marcel Dekker, Inc. Marcel Dekker, Inc. grants you the limited right to display the Materials only on your personal computer or personal wireless device, and to copy and download single copies of such Materials provided that any copyright, trademark or other notice appearing on such Materials is also retained by, displayed, copied or downloaded as part of the Materials and is not removed or obscured, and provided you do not edit, modify, alter or enhance the Materials. Please refer to our Website User Agreement for more details.

Order now!

Reprints of this article can also be ordered at http://www.dekker.com/servlet/product/DOI/101081NCN100002419